





## A facile synthesis of 3-alkoxy and 3-amino pyrroles

Valérie Breuil-Desvergnes, Philippe Compain, <sup>†</sup> Jean-Michel Vatèle and Jacques Goré \*

Laboratoire de Chimie Organique I, associé au CNRS, Université Claude Bernard CPE-Lyon,

43 boulevard du 11 novembre 1918, 69622 Villeurbanne, France

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## Abstract

N-Amino-3-methoxy-3-pyrrolines 2 obtained from the reaction of 1-lithio methoxyallene with arylhydrazones may be converted to either 3-methoxy pyrroles 6 or 3-amino pyrroles 7 by treatment with m-chloroperbenzoic acid or 0.25 M HCl, respectively. © 1999 Elsevier Science Ltd. All rights reserved.

We recently reported that the lithio-derivative of methoxyallene reacted with the SAMP-hydrazones 1a to afford 3-pyrrolinohydrazines 2a. This cyclisation was easily accomplished in one step for arylhydrazones, or in two steps for alkyl and arylhydrazones (Scheme 1).

Scheme 1.

In order to extend the scope and synthetic utility of this approach, the reactions of other hydrazones under similar conditions have been studied. Dimethylhydrazones, pyrrolydinyl hydrazones and morpholinyl hydrazones were easily synthesised from aryl and alkyl aldehydes and commercially available hydrazines. The condensation reaction was then performed under the conditions employed previously, using 6 equiv. of organometallic reagent in THF. The cyclisation of aryl hydrazones proved more difficult

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<sup>\*</sup> Corresponding author. Tel: 33-04-72-44-81-35; fax: 33-04-72-43-12-14; e-mail: gore@univ-lyon1.fr

<sup>†</sup> Present address: Institut de Chimie Organique et Analytique, ESA 6005, BP 6759 45067 Orléans Cédex 2, France.

than that of the previously reported SAMP-hydrazones. Addition at  $-78^{\circ}$ C of the organometallic reagent afforded the desired cyclised products after warming to room temperature (rt) for several hours ( $-20^{\circ}$ C in the case of 1a). The 3-pyrrolinohydrazines 2b-d were not the sole reaction products, however, and obtention of the four-membered cycles 4b-d was also observed (Scheme 2). The ratio of four- to five-membered ring formation was found to depend on the nature of the substituents on the terminal nitrogen atom. Formation of the four-membered rings was favoured by N,N-dimethyl and piperidinyl hydrazones over their morpholinyl analogues; in this latter case, formation of the four-membered ring may be suppressed by performing the reaction at  $50^{\circ}$ C (see Table 1).

Scheme 2.

In contrast, the corresponding alkylhydrazones (R=propyl, isopropyl) failed to yield any cyclised product. Neither direct cyclisation in THF nor cyclisation of the allenic hydrazines **3b-d** resulting from reaction in diethyl ether of 1-lithio methoxyallene with hydrazones proved successful.

This approach has provided a route to diversely substituted 3-pyrrolinyl hydrazines 2. In the case of the SAMP-series, 3-pyrrolines 5 may be obtained as a single enantiomer via hydrogenolysis of these 3Table 1

Starting hydrazone		Reaction conditions		Reaction product	
R,R	Ar	Temperature	Time	2 (yield%)	4 (yield%)
	1ba Ph	RT	16h	2ba 53	4ba 22
		50°C	6h	46	26
Ме	1bb p-MePh	RT	16h	2bb 45	4bb 35
1b	1bc p-MeOPh	RT	16h	2be 35	4bc 38
	1bd β-naphthyl	RT	16h	2bd 77	•
	1ca Ph	RT	16h	2ca 46	4ca 28
-(CH <sub>2</sub> ) <sub>5</sub> -		RT	96h	46	27
piperidino	1cb p-MeOPh	RT	16h	2cb 38	4cb 38
1e	1cc β-naphthyl	RT	16h	2ec 78	4cc 4
-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> -	1da Ph	RT	16h	2da 64	4da 11
		50°C	6h	74	•
	1db p-MeOPh	50°C	6h	2db 76	•
morpholino	1de β-naphthyl	RT	16h	2de 60	4dc 13
1d		50°C	6h	78	•
	1dd p-MePh	50°C	6h	2dd 78	-

In series b and d, the products 2 and 4 are separated by SG-Flash chromatography. The yields correspond to quantities of isolated pure compounds. In sery 1c, the same separation was not possible; the yields are based on the integration of the <sup>1</sup>H NMR spectra and on the quantity of the purified mixture of 2 and 4.

pyrrolinyl hydrazines 2.<sup>1</sup> We wish now to report the transformation of the same compounds into the pyrroles 6 and 7. Although a plethora of methods are available for the preparation of pyrroles, this work most closely resembles those routes in which the final step is conversion of a 3-pyrroline into a pyrrole.<sup>2</sup> The advantage of this new method is in allowing the synthesis of pyrroles functionalised with electrodonating group in the 3-position, for which relatively few examples exist in the literature.<sup>3</sup>

The first transformation was run with the N-morpholino-3-pyrrolines 2d which were obtained with good yields in a single step (Table 1). Treatment of N-morpholino-3-pyrroline 2d with m-CPBA in dichloromethane at 25°C afforded the pyrrole 6 in good yield (Scheme 3). This reaction is thought to involve a Cope elimination of an N-oxide followed by tautomerisation of the heterocycle.<sup>4</sup> It is worth noting that no reaction was observed when  $H_2O_2$  was used in place of m-CPBA.

Scheme 3.

On the other hand, treatment of the compounds 2 with dilute HCl (0.25 M) in THF (1:4) resulted in the somewhat surprising transformation of 2 into 7. The 1,3-migration of the external amino group occurs quantitatively regardless of the nature of the substituents: SMP, dimethylamino, piperidino or morpholino (Scheme 4).

Scheme 4.

The mechanism of this migration probably involves the formation of a bridged ammonium species  $8^5$  followed by sequential  $\beta$ -elimination and elimination of methanol to give 7 (Scheme 5).

Examples of 3-amino pyrroles obtained are shown (Scheme 6). With 2a, 2b and 2d the corresponding 3-amino pyrrole was obtained cleanly in quantitative yield. In the case of 2c, where separation of the four-and five-membered cycles 2ca and 4ca proved problematic, the reaction was performed on a mixture of the two where  $R=\beta$ -naphthyl. However, the 3-amino pyrrole 7cc was the only product isolated from the

Scheme 5.

reaction, suggesting that the four-membered cycle 4cc is degraded under the reaction conditions. The 3-amino pyrroles 7 are relatively unstable to chromatography but the crude product presents a  $\geq$ 95% purity as shown by  $^{1}$ H and  $^{13}$ C NMR spectroscopy.

Scheme 6.

In conclusion, a two step preparation of 3-methoxy and 3-dialkylamino pyrroles has been developed. The method extends the synthetic utility of alkoxyallenes, with the potential for the scope of this reaction to be further applied to other alkoxy and amino groups.

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